

5-Acetyl-4-methyl-2-(*o*-toluidinyl)-1,3-thiazoleBohari M. Yamin,* Noor Azilah
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Key indicators

Single-crystal X-ray study
 $T = 273$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.044
 wR factor = 0.122
Data-to-parameter ratio = 16.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the title compound, $\text{C}_{13}\text{H}_{14}\text{N}_2\text{OS}$, the benzene and thiazole rings make a dihedral angle of 73.44 (10)°. The intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds link the molecules into centrosymmetric dimers. The crystal packing is further stabilized by van der Waals forces.

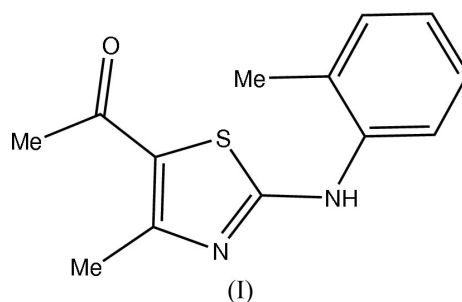
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Comment

In the title molecule, (I) (Fig. 1), all bond lengths and angles are in normal ranges (Allen *et al.*, 1987). The geometry of the thiazole ring (Table 1) is close to that observed in its trisubstituted analogue methyl 2-amino-isopropyl-1,3-thiazole-4-carboxylate (Kennedy *et al.*, 2004). The $\text{S1}/\text{C8}/\text{N2}/\text{C9}/\text{C10}/\text{N1}/\text{C13}$ moiety and the $\text{C11}/\text{C12}/\text{O1}$ acetyl group are essentially coplanar, with a maximum deviation of 0.009 (2) Å for atom N1. The dihedral angle between the thiazole ring and the acetyl fragment is 9.15 (14)°. The benzene and thiazole rings make a dihedral angle of 73.44 (10)°. In the molecule, there is a weak intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond (Table 2). Intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds (Table 2) link the molecules into centrosymmetric dimers, arranged parallel to the *ac* face. The crystal packing (Fig. 2) is further stabilized by van der Waals forces.



Experimental

A solution of *o*-toluidine (1.34 g, 0.01 mol) in acetone (50 ml) was added dropwise to an acetone solution (50 ml) containing an equimolar amount of 3-chloroacetylacetone and ammonium thiocyanate in a two-necked round-bottomed flask. The solution was refluxed for about 1 h. The light-yellow solution was filtered and some colourless crystals were obtained after 5 d of evaporation (yield 80%, m.p. 442.8 – 442.1 K). ^1H NMR: δ 2.34 (3H, s, H-7), 2.46 (3H, s, H-13), 2.40 (3H, s, H-12), 10.10 (H, s, NH-2), 7.22–7.50 (CH aromatic, 2–5); ^{13}C NMR: δ 17.8 (C13), 18.3 (C7), 29.9 (C12), 131.6 (C1), 133.6 (C2), 125.3 (C3), 127.6 (C4), 121.5 (C5), 137.8 (C6), 157.5 (C11), 171.5 (C9), 189.3 (C8).

Crystal data

$C_{13}H_{14}N_2OS$
 $M_r = 246.32$
 Monoclinic, $P2_1/c$
 $a = 10.797$ (2) Å
 $b = 11.571$ (2) Å
 $c = 12.0982$ (16) Å
 $\beta = 123.576$ (11)°
 $V = 1259.3$ (4) Å³
 $Z = 4$

$D_x = 1.299$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 871 reflections
 $\theta = 2.2$ – 26.5°
 $\mu = 0.24$ mm⁻¹
 $T = 273$ (2) K
 Block, colourless
 $0.41 \times 0.40 \times 0.33$ mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.907$, $T_{\max} = 0.924$
 6884 measured reflections

2610 independent reflections
 2343 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$
 $\theta_{\text{max}} = 26.5^\circ$
 $h = -13 \rightarrow 8$
 $k = -14 \rightarrow 14$
 $l = -14 \rightarrow 15$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.122$
 $S = 1.08$
 2610 reflections
 157 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0659P)^2 + 0.3401P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.28$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.22$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

S1—C8	1.7269 (16)	N2—C8	1.317 (2)
S1—C9	1.7527 (19)	N2—C10	1.368 (2)
O1—C11	1.219 (3)	C9—C10	1.370 (2)
N1—C8	1.349 (2)		
C8—S1—C9	88.65 (8)	C9—C10—C13	126.24 (17)
C10—C9—C11	128.87 (18)		

Table 2

Hydrogen-bonding geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C13—H13B \cdots O1	0.96	2.50	2.982 (4)	111
N1—H1A \cdots N2 ⁱ	0.86	2.17	2.958 (3)	153

Symmetry code: (i) $-x, -y, -z$.

After their location in a difference Fourier map, all H atoms were positioned geometrically and allowed to ride on the parent C or N atoms, with $C-H = 0.93$ – 0.96 Å, $N-H = 0.86$ Å and $U_{\text{iso}} = 1.2$ – $1.5U_{\text{eq}}$ (parent atom).

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL (Sheldrick, 1997); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

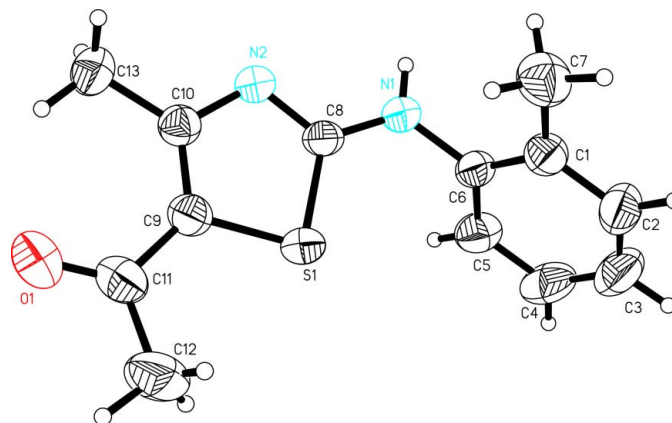


Figure 1
View of (I) with 50% probability displacement ellipsoids.

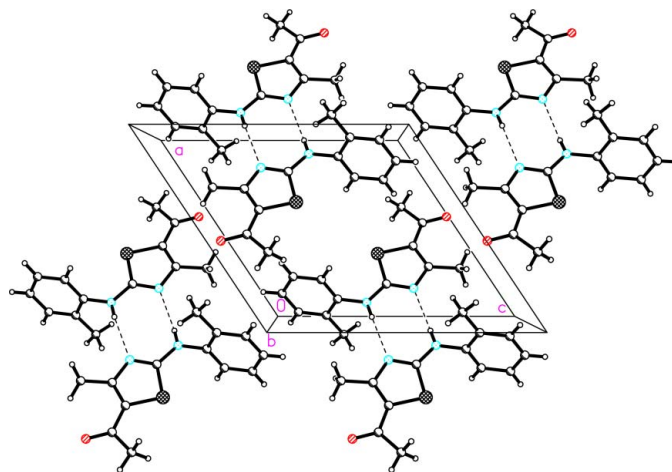


Figure 2
The packing, viewed down the b axis. Dashed lines denote intermolecular $N-H\cdots N$ hydrogen bonds.

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Kennedy, A. R., Khalaf, A. I., Suckling, C. J. & Waigh, R. D. (2004). *Acta Cryst. E* **60**, o1510–o1512.
 Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
 Siemens (1996). *SMART* and *SAINTE*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.